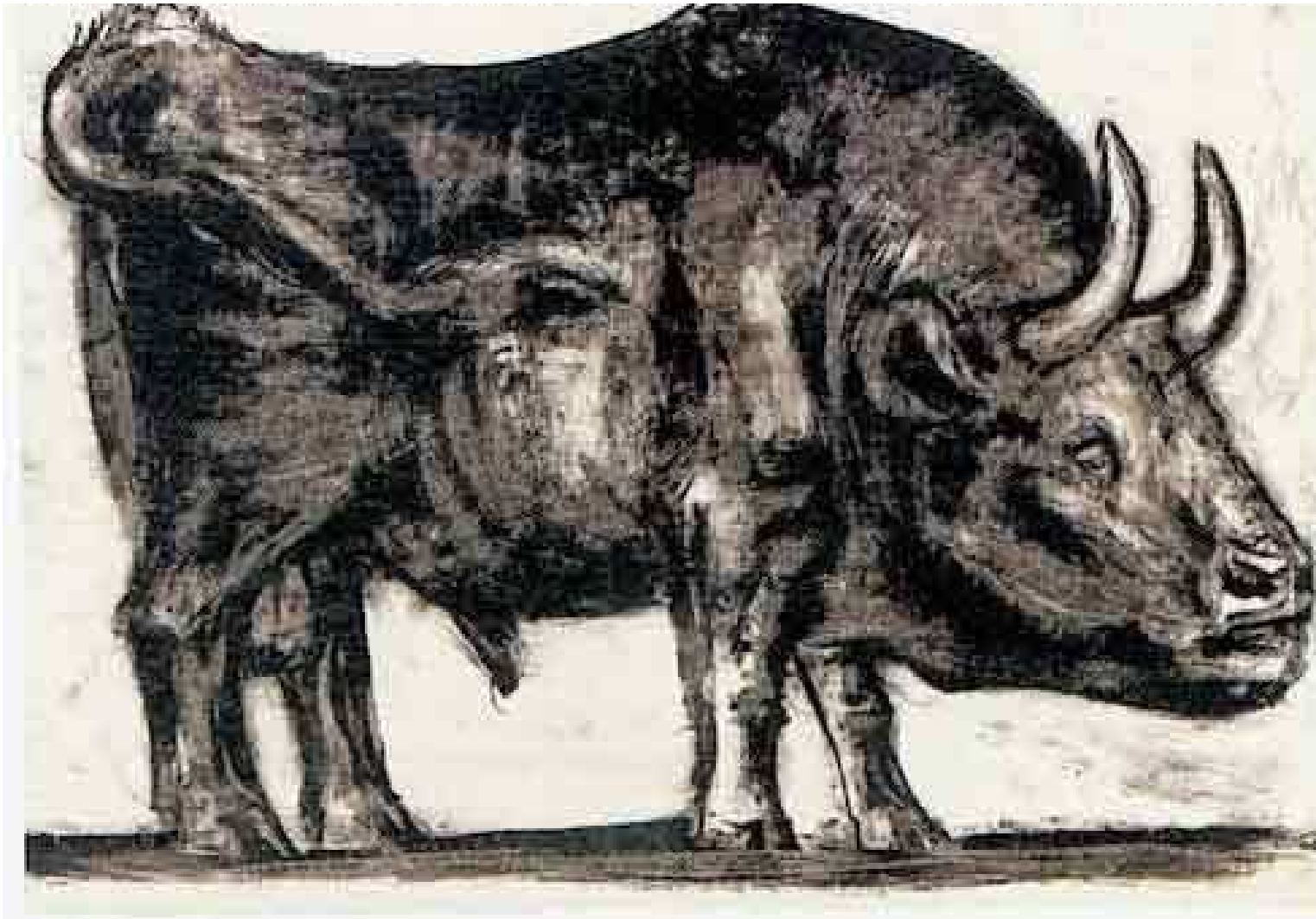


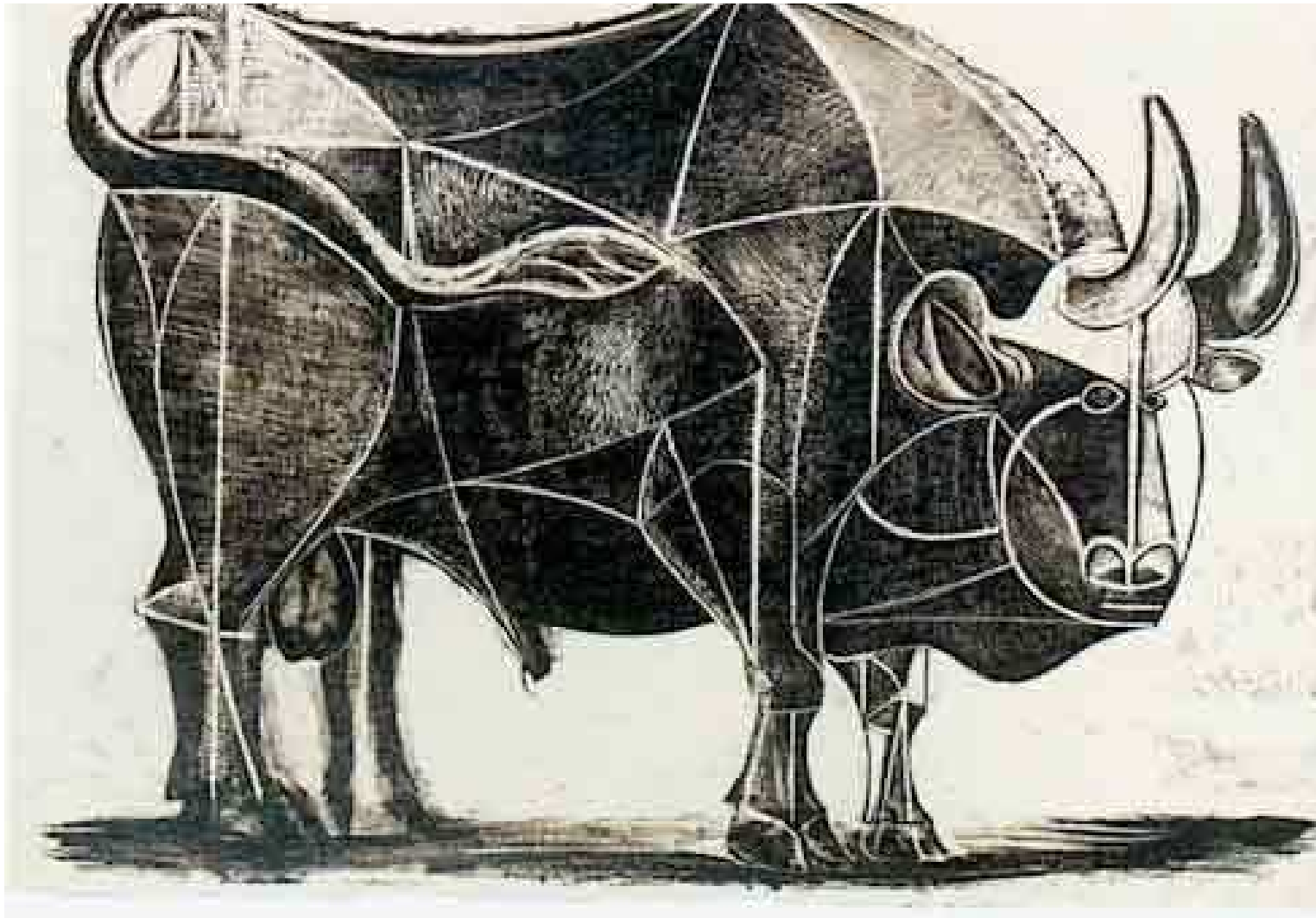
Demarche Scientifique



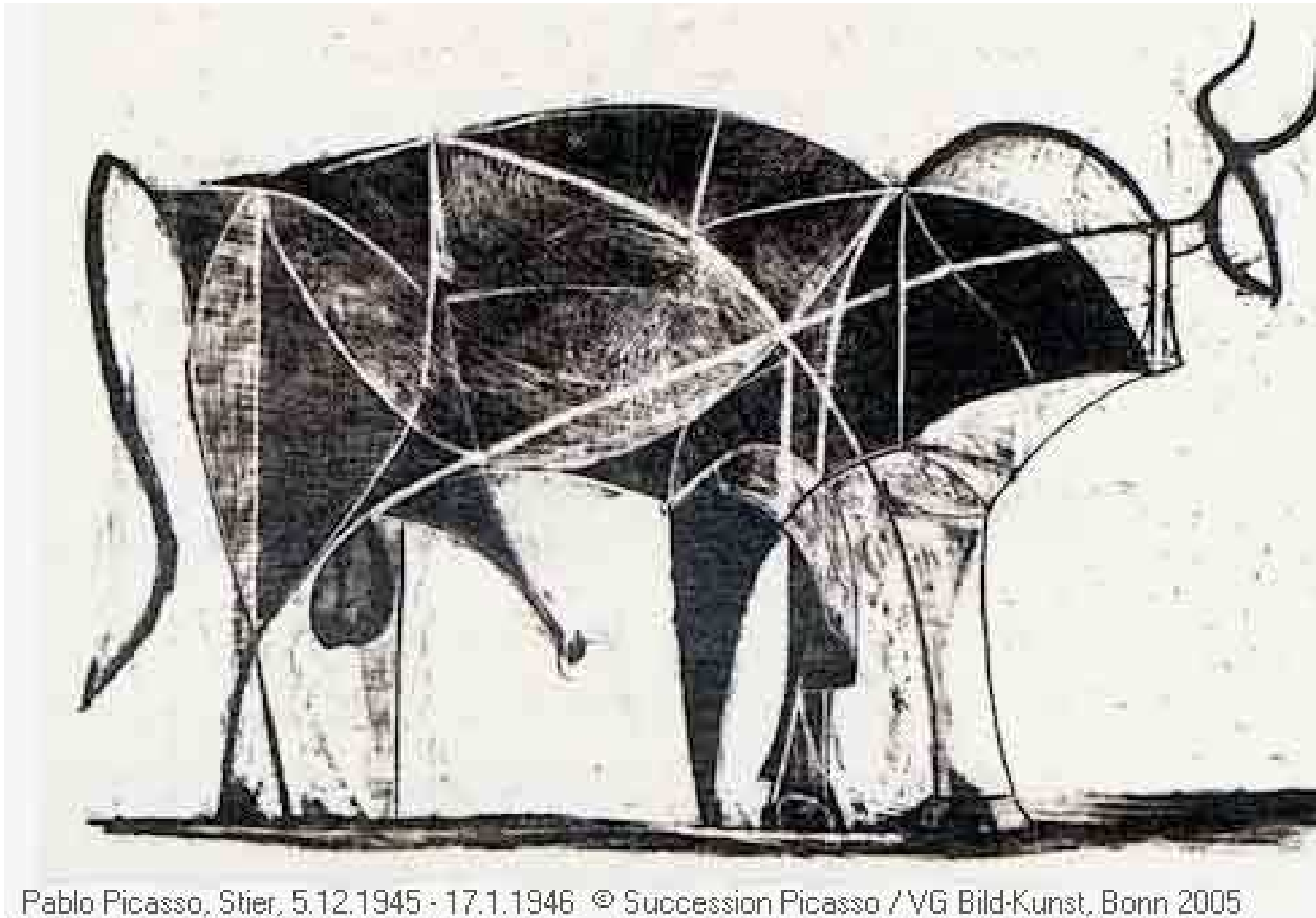
Demarche Scientifique



Demarche Scientifique

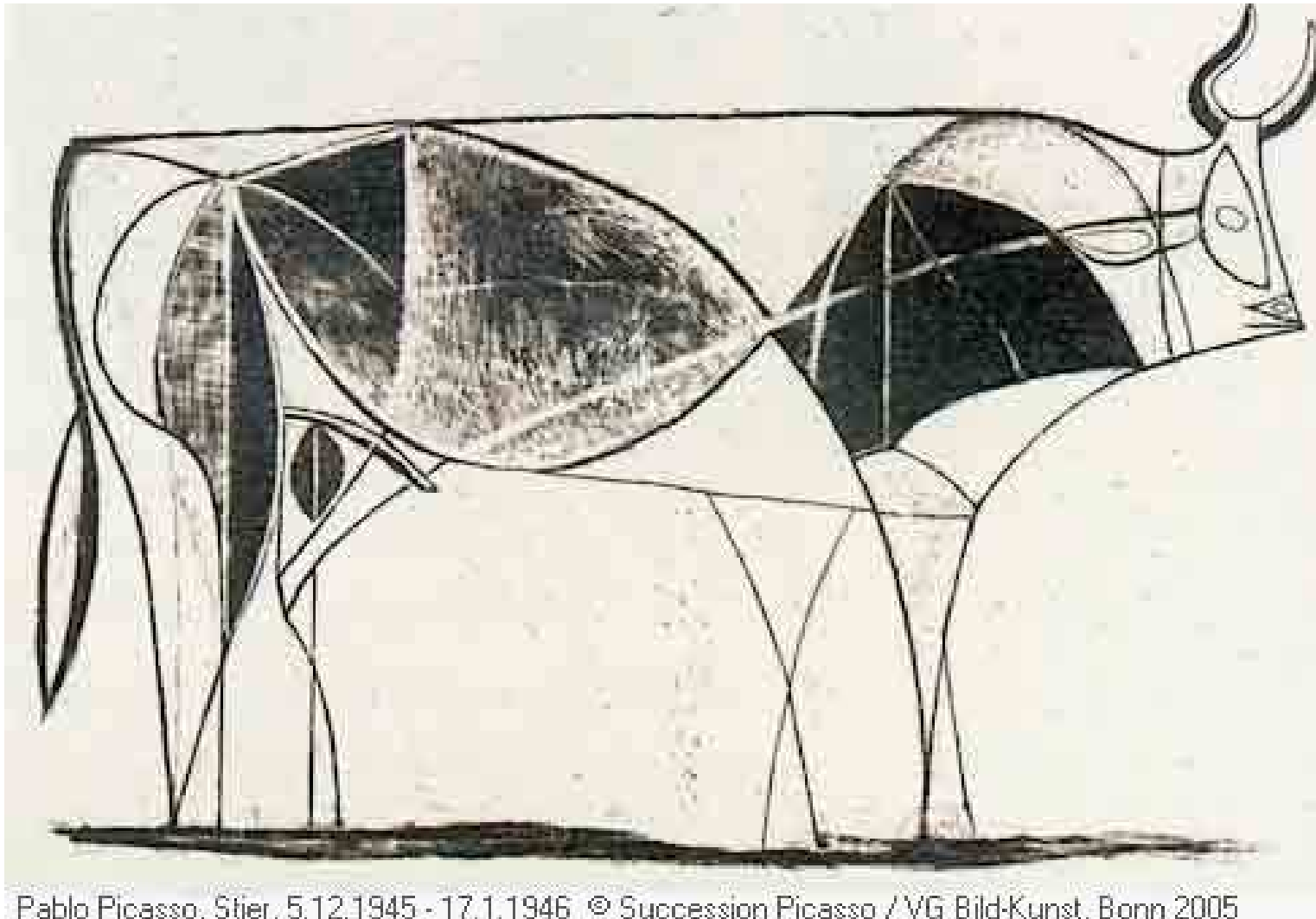


Demarche Scientifique



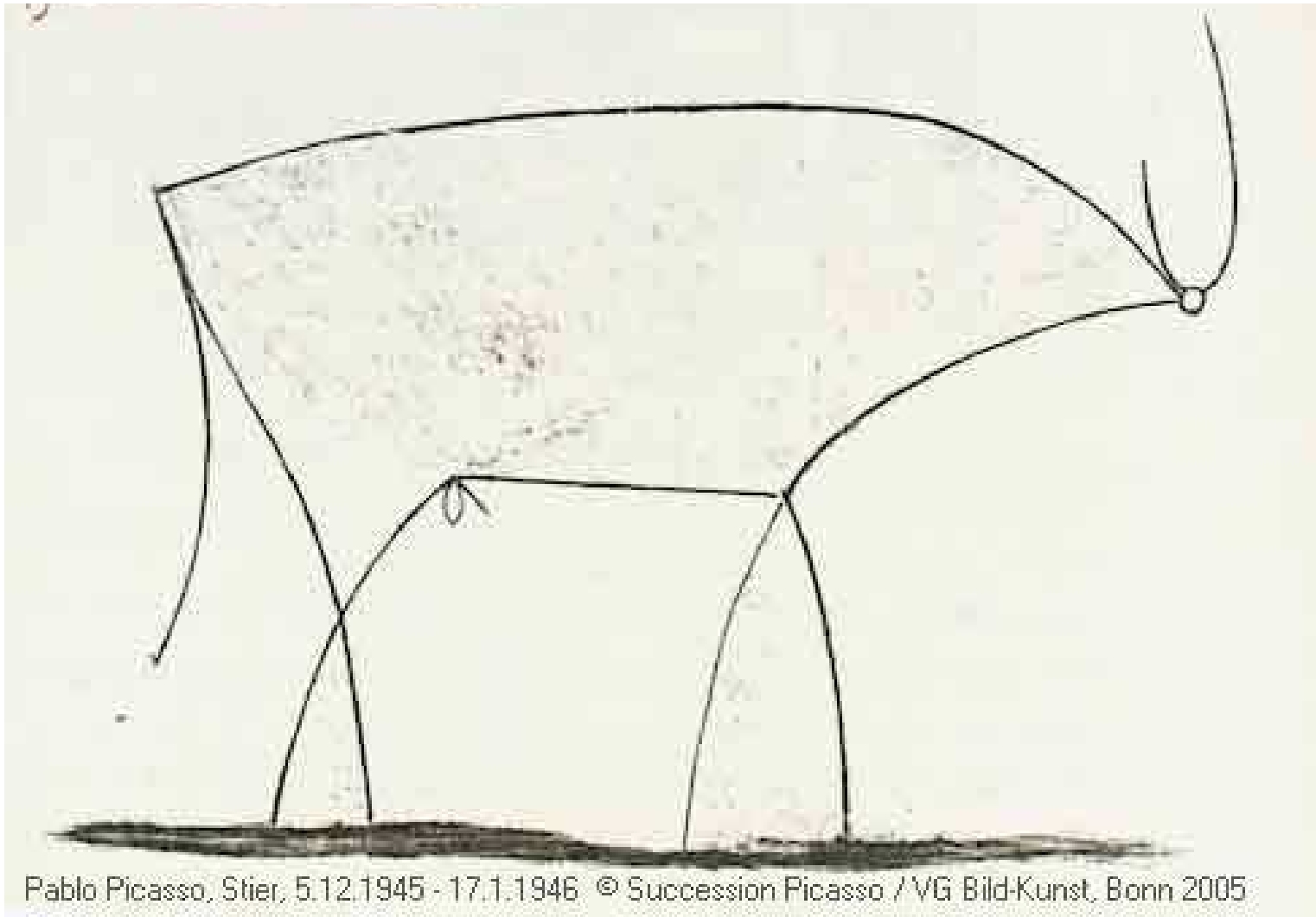
Pablo Picasso, Stier, 5.12.1945 - 17.1.1946 © Succession Picasso / VG Bild-Kunst, Bonn 2005

Demarche Scientifique

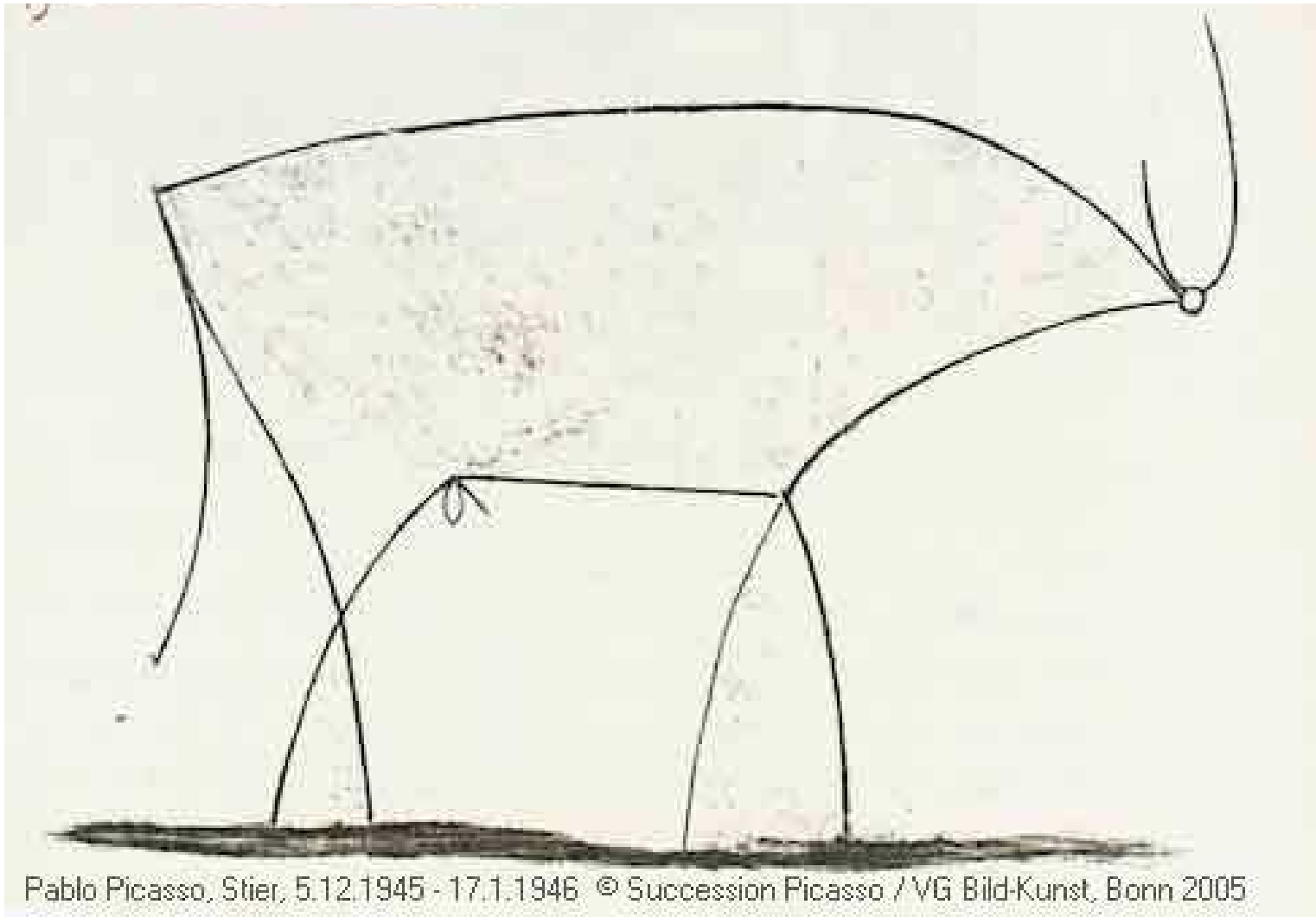


Pablo Picasso. Stier. 5.12.1945 - 17.1.1946 © Succession Picasso / VG Bild-Kunst, Bonn 2005

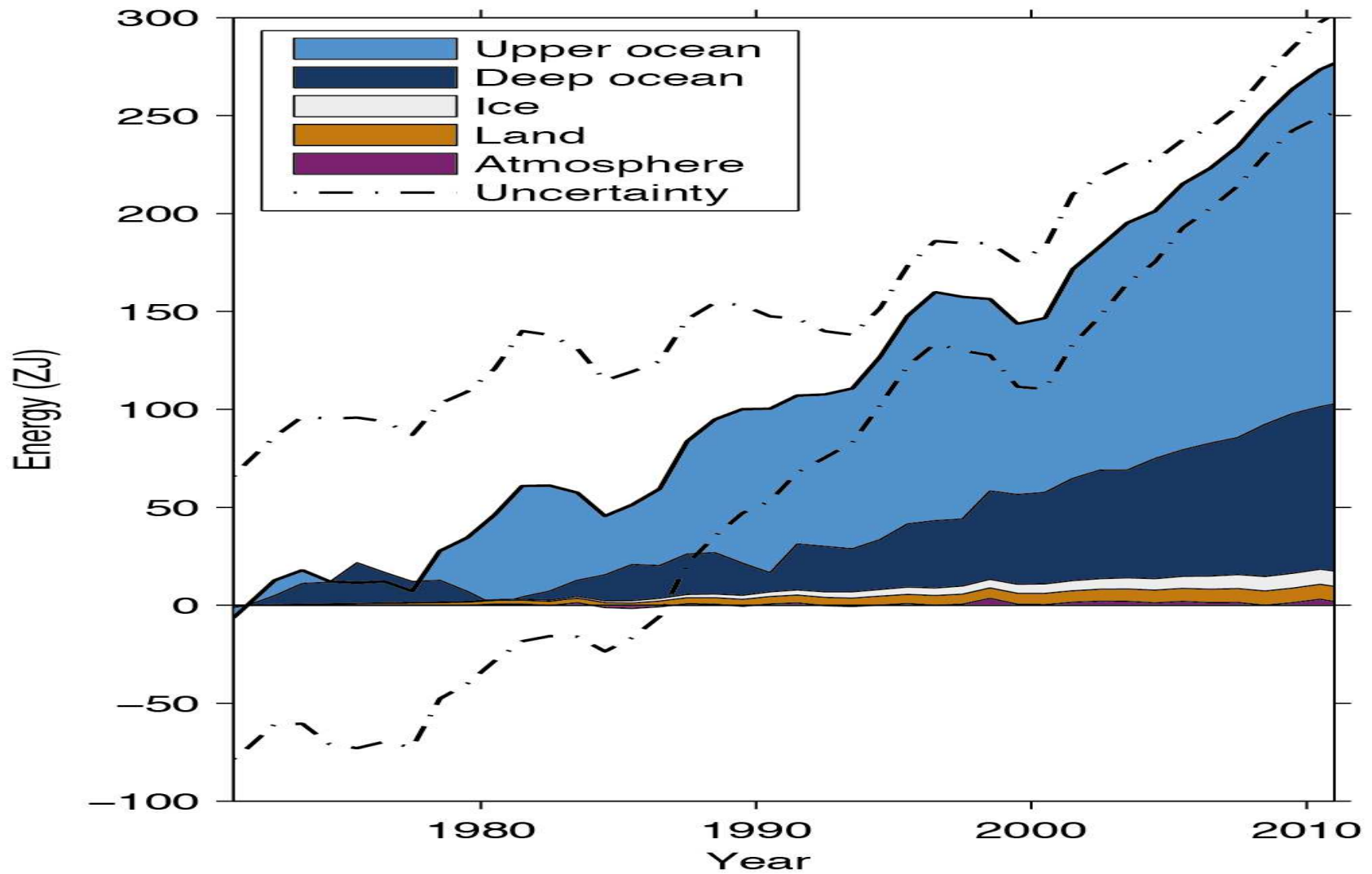
Demarche Scientifique



Demarche Scientifique

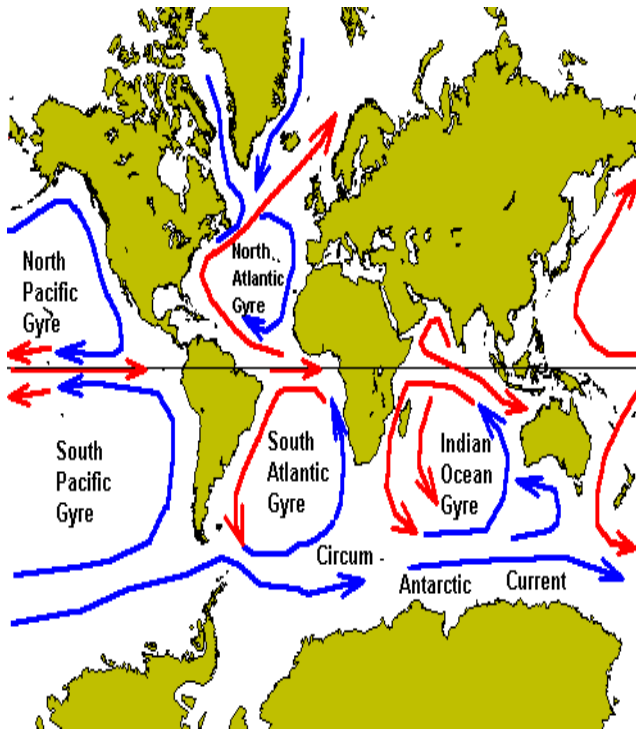


Heat content (IPCC)

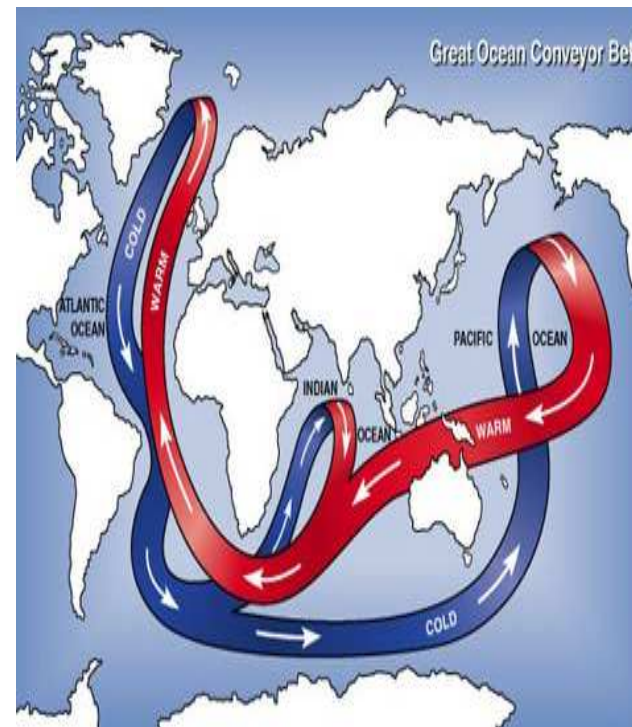


Ocean Circulation

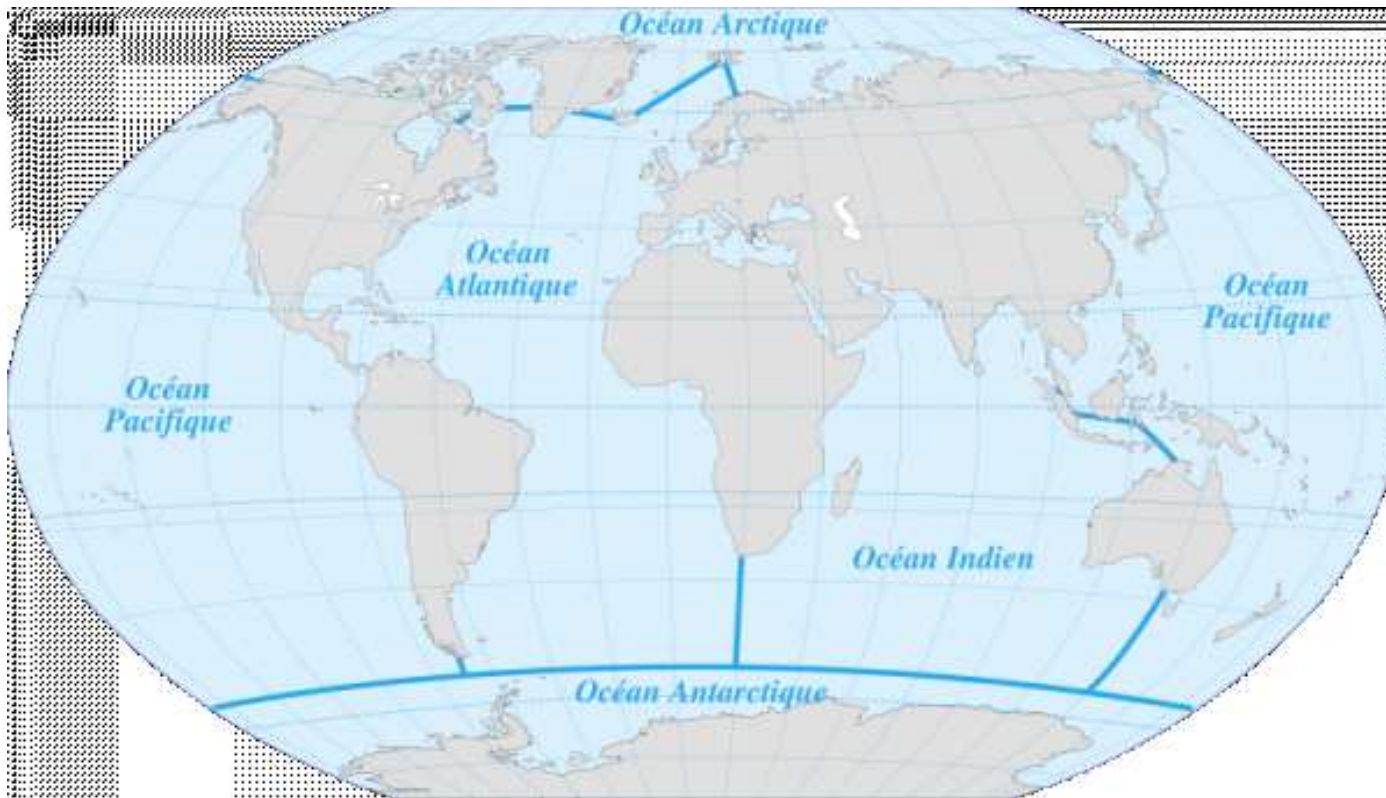
Gyre
“weather”



Overturning
“climat”



Les Océans

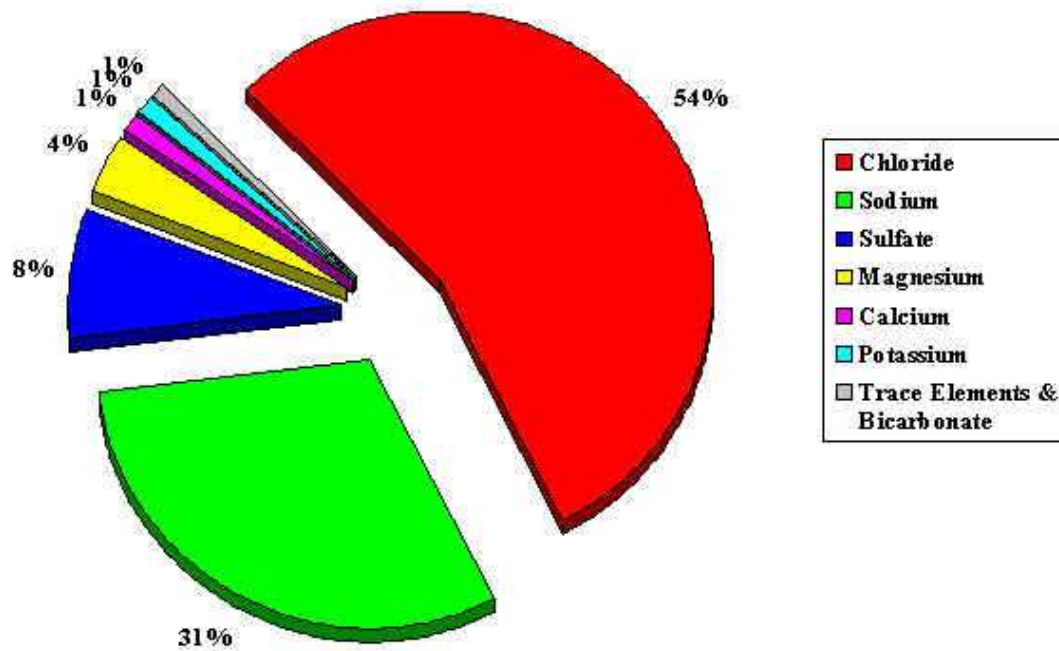


H.M.S. Challenger 1872 – 1876



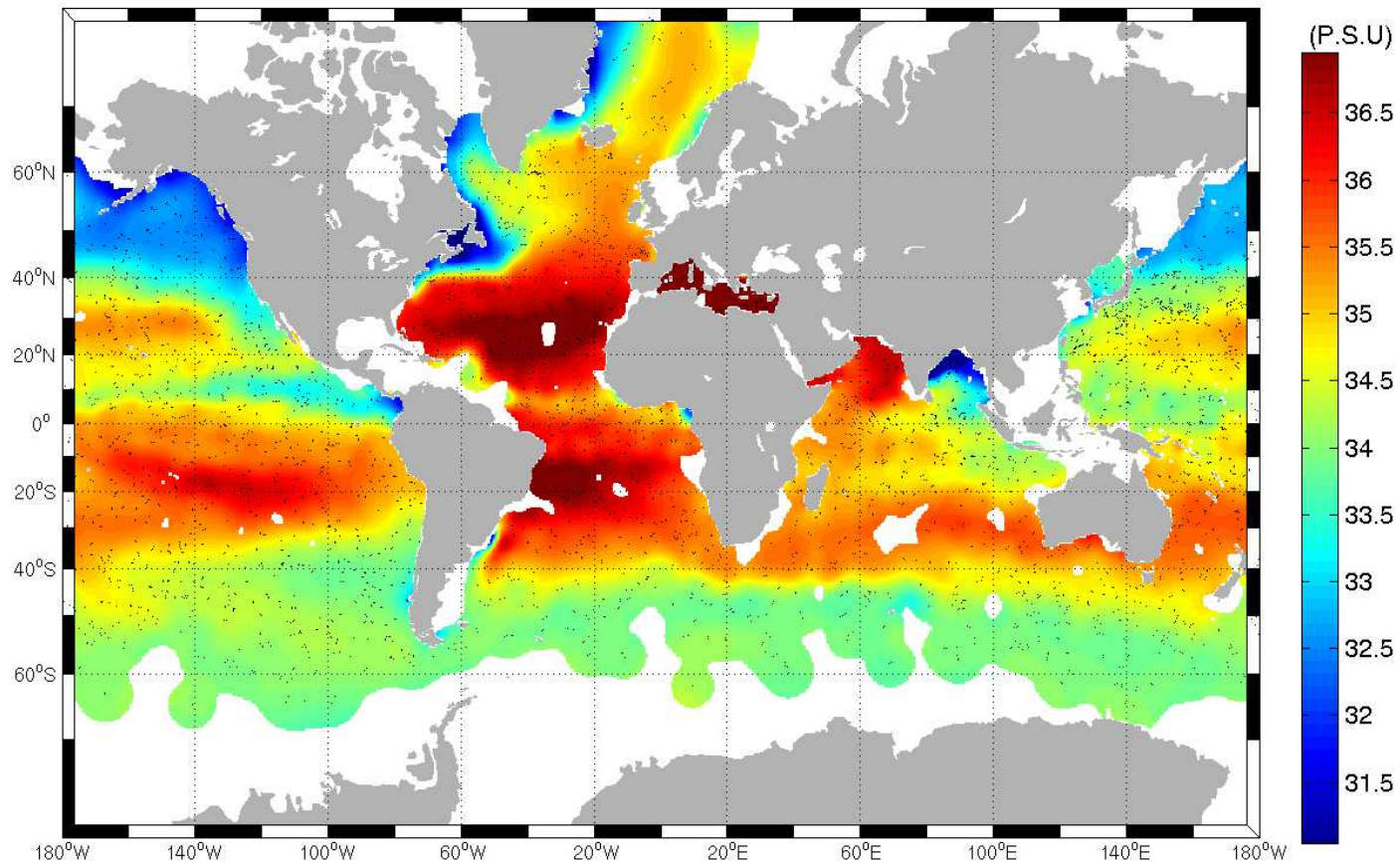
Principe de Marcet ; Loi de Dittmar

Constituents of Seawater



Salinité à 10m

Salinity analysis (P.S.U) – Depth 10 m – 30-Aug-2006

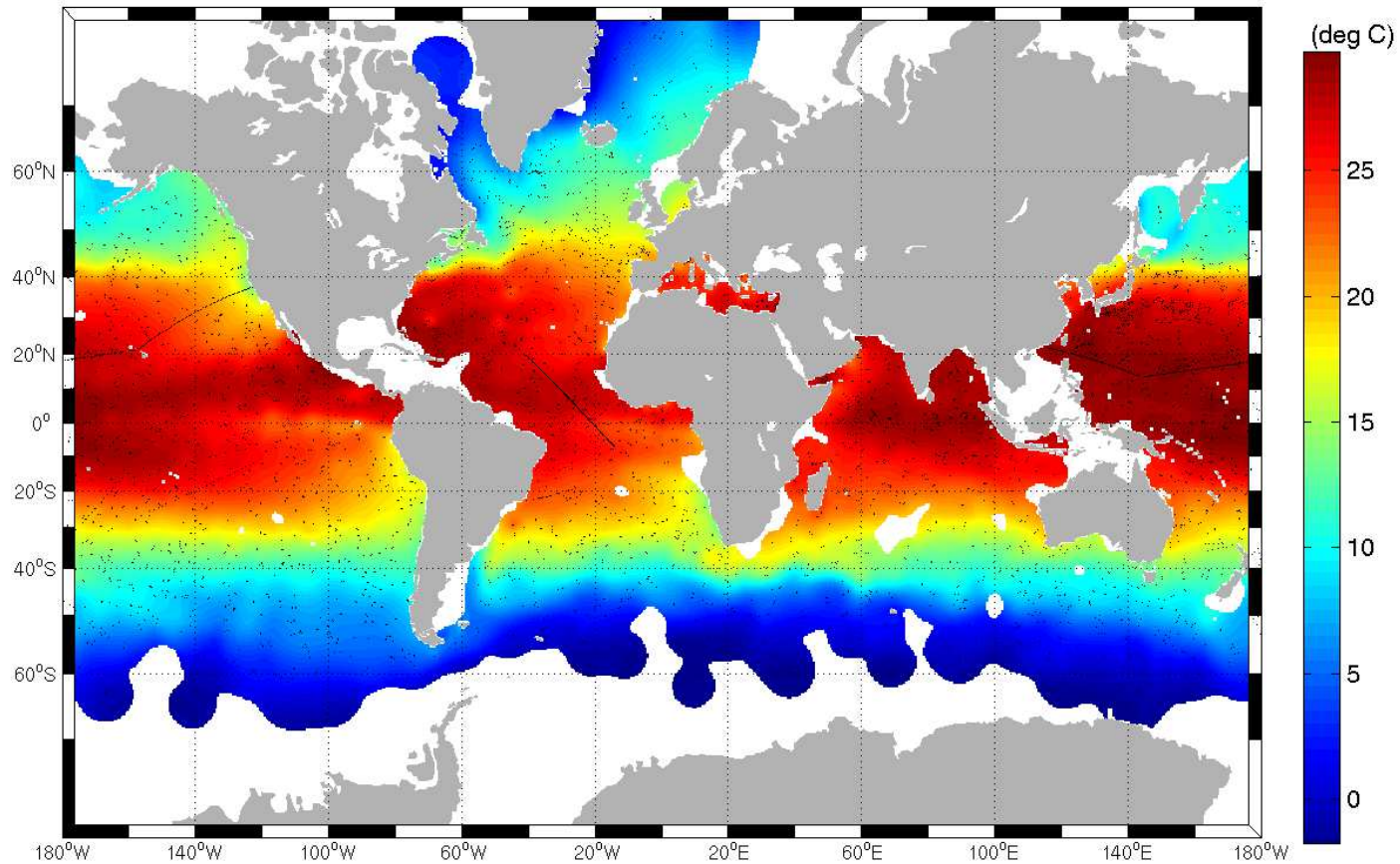


Coriolis

min = 18.17 max = 39.58 Last update : 30-Aug-2006

Temp. à 10m

Temperature analysis (deg C) – Depth 10 m – 30-Aug-2006



Coriolis

min = -2.05 max = 30.44 Last update : 30-Aug-2006

Temp. vs Temp. potentielle

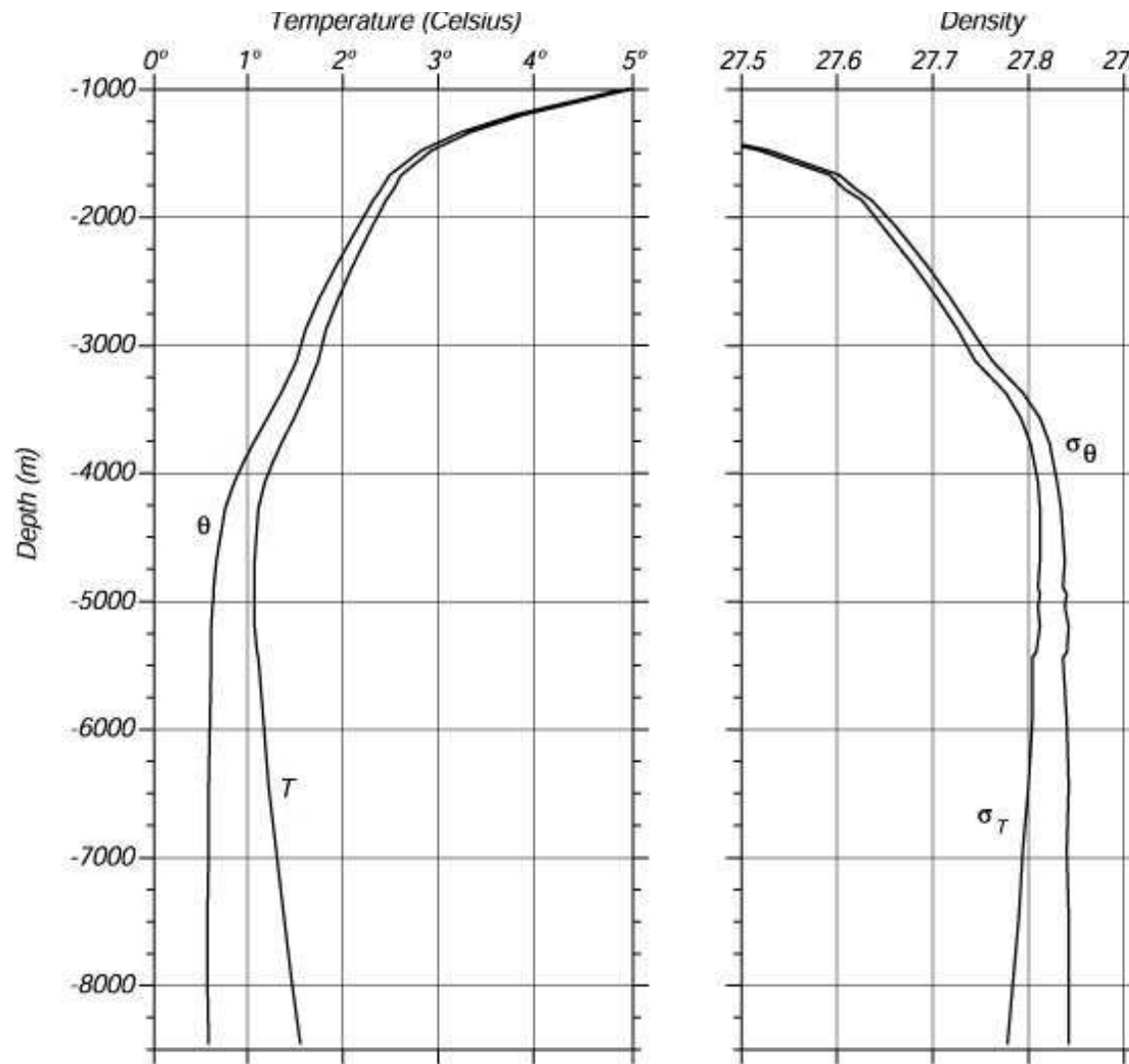


Diagram TS

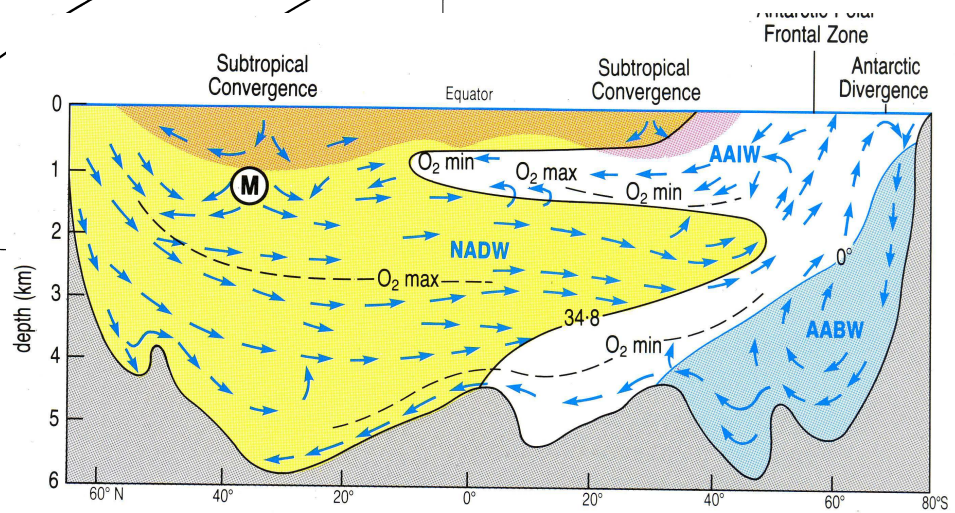
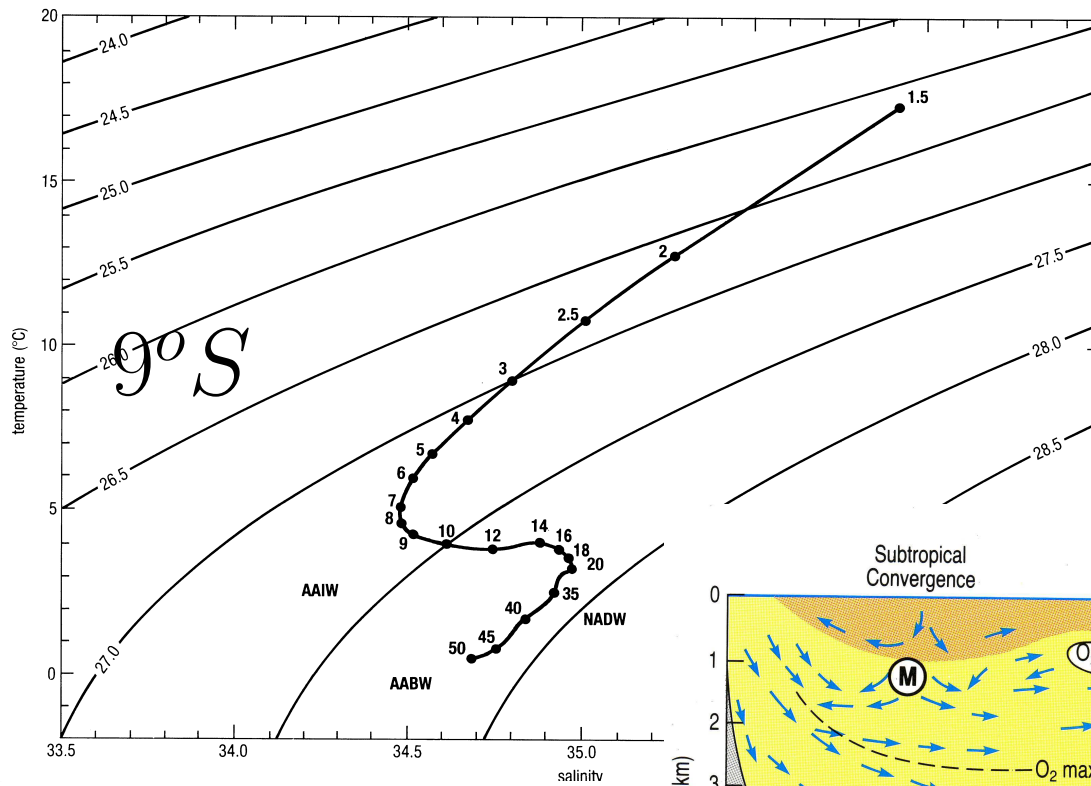


Diagram TS

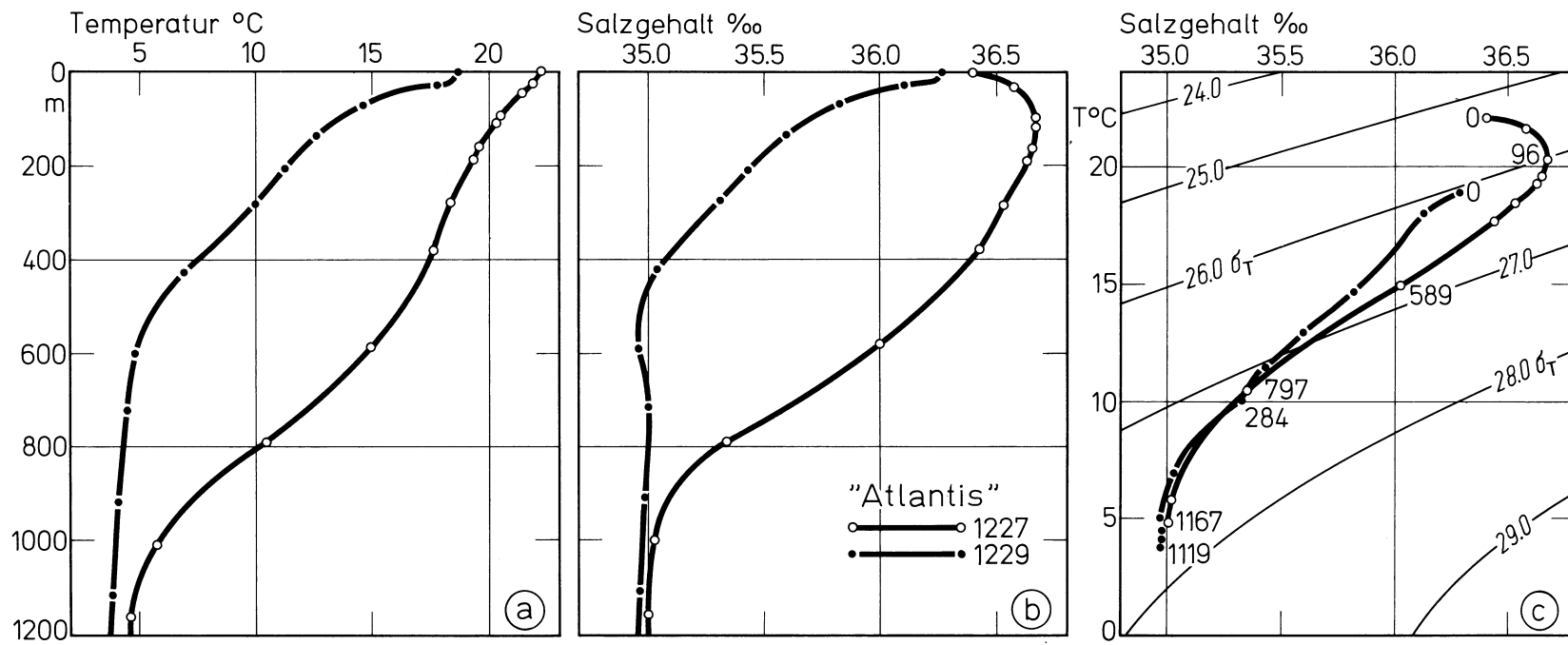


Abb. 5.16. Beziehung zwischen Temperatur und Salzgehalt auf den beiden benachbarten Stationen

Diagram TS-Volumique

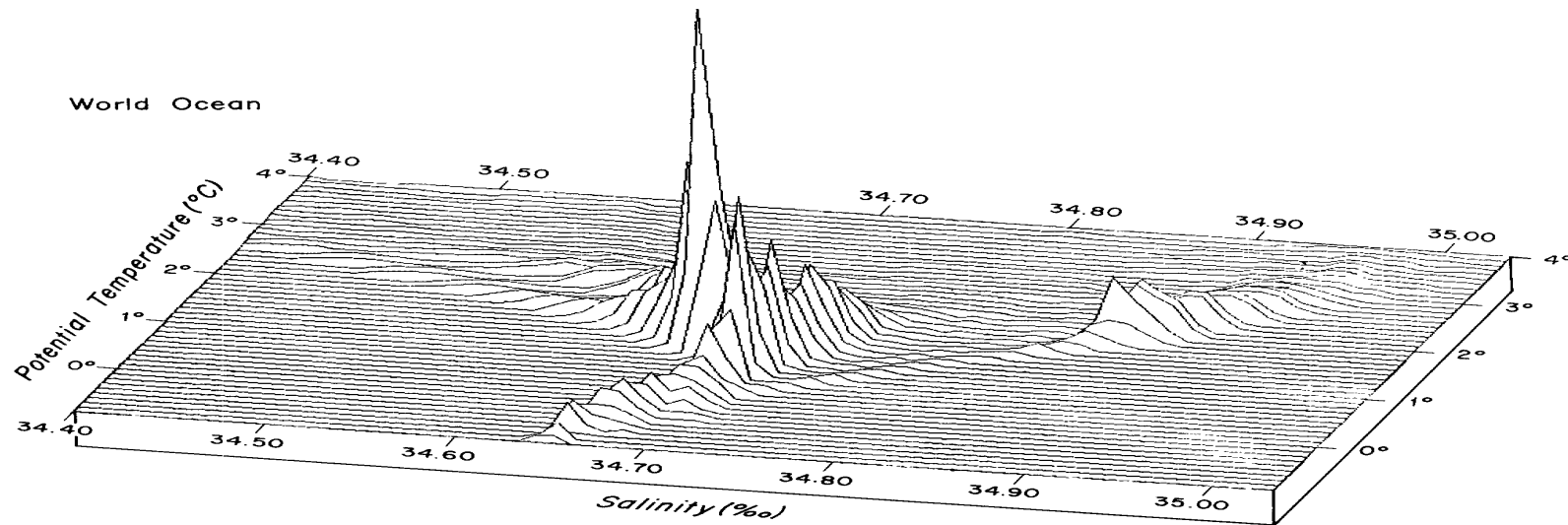
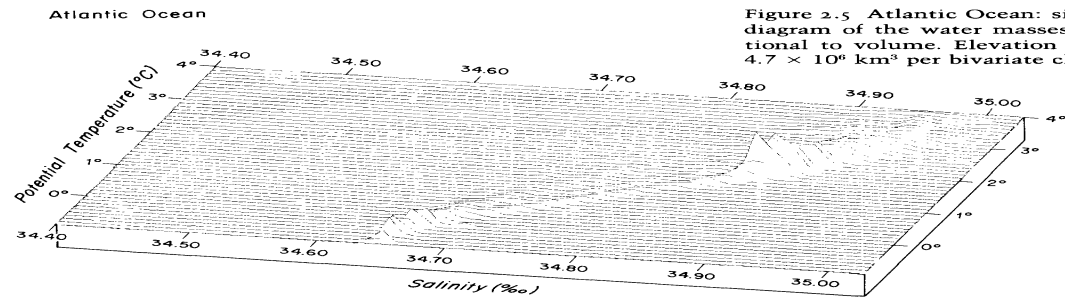
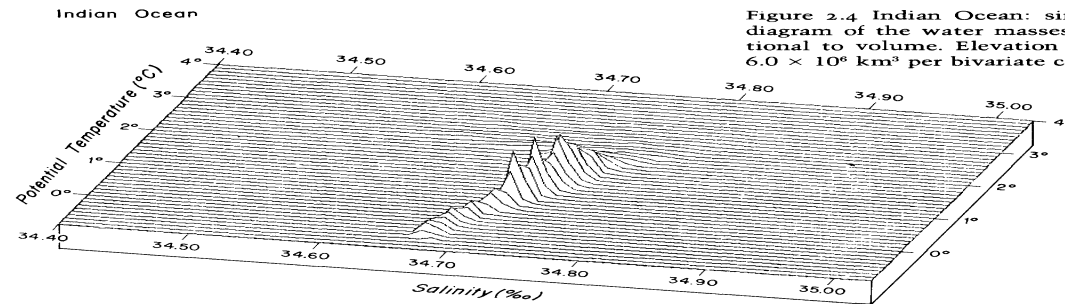
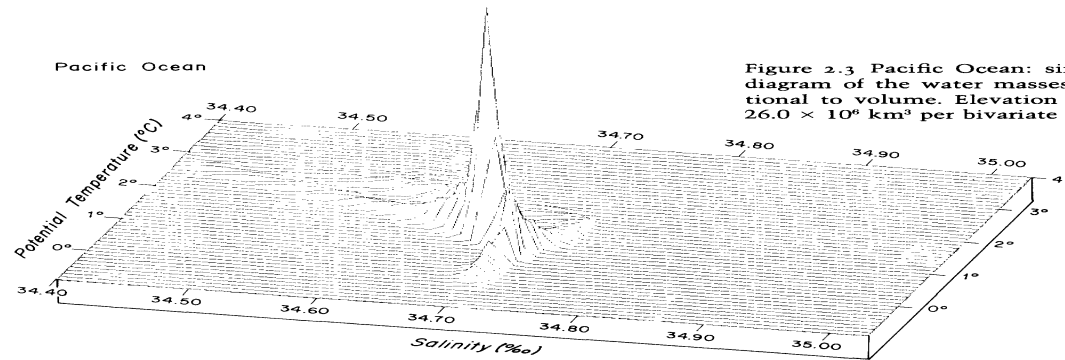


Diagram TS-Volumique



Densité et Congélation

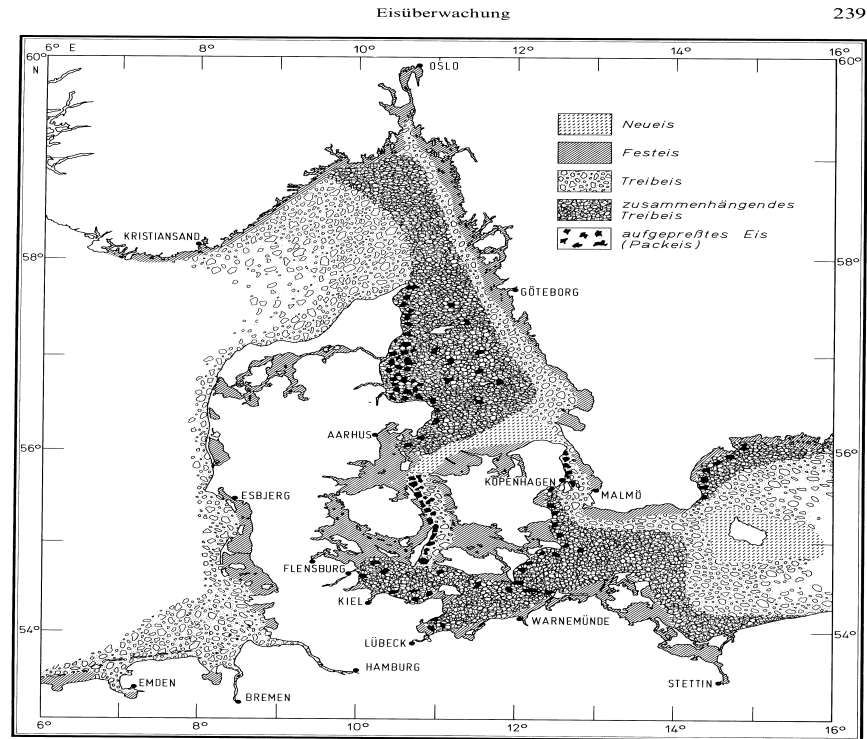
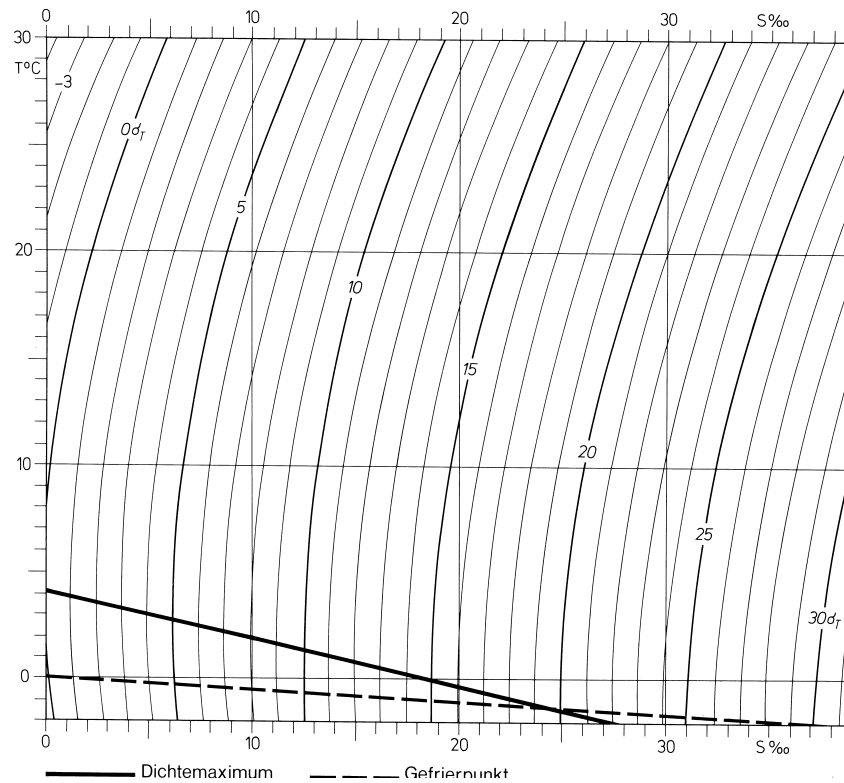


Abb. 5.32. Eisverhältnisse in der östlichen Nordsee und westlichen Ostsee am 20. Februar 1963 (nach Eisübersichtskarte des Deutschen Hydrographischen Instituts, Hamburg). Stellt maximale Eisbedeckung im Zeitraum 1896-1971 dar.

Equation d'état (UNESCO)

The One Atmosphere International Equation of State of Seawater, 1980

Definition

The density (ρ , kg m⁻³) of seawater at one standard atmosphere ($p = 0$) is to be computed from the practical salinity (S) and the temperature (t , °C) with the following equation :

$$\begin{aligned}\rho(S, t, 0) = & \rho_w + (8.244\ 93 \times 10^{-1} - 4.0899 \times 10^{-3} t \\ & + 7.6438 \times 10^{-5} t^2 - 8.2467 \times 10^{-7} t^3 + 5.3875 \times 10^{-9} t^4) S \\ & + (-5.724\ 66 \times 10^{-3} + 1.0227 \times 10^{-4} t - 1.6546 \times 10^{-6} t^2) S^{3/2} \\ & + 4.8314 \times 10^{-4} S^2\end{aligned}$$

where ρ_w , the density of the Standard Mean Ocean Water (SMOW) taken as pure water reference, is given by

$$\begin{aligned}\rho_w = & 999.842\ 594 + 6.793\ 952 \times 10^{-2} t - 9.095\ 290 \times 10^{-3} t^2 \\ & + 1.001\ 685 \times 10^{-4} t^3 - 1.120\ 083 \times 10^{-6} t^4 \\ & + 6.536\ 332 \times 10^{-9} t^5\end{aligned}$$

The one atmosphere International Equation of State of Seawater, 1980 is valid for practical salinity from 0 to 42 and temperature from -2 to 40°C.

The High Pressure International Equation of State of Seawater, 1980

Definition

The density (ρ , kg m⁻³) of seawater at high pressure is to be computed from the practical salinity (S), the temperature (t , °C) and the applied pressure (p , bars) with the following equation :

$$\rho(S, t, p) = \frac{\rho(S, t, 0)}{1 - p/K(S, t, p)}$$

where $\rho(S, t, 0)$ is the one atmosphere International Equation of State 1980, given on the preceding front page and $K(S, t, p)$ is the secant bulk modulus given by

$$K(S, t, p) = K(S, t, 0) + A p + B p^2$$

where

$$\begin{aligned}K(S, t, 0) = & K_w + (54.6746 - 0.503\ 459 t + 1.099\ 87 \times 10^{-2} t^2 \\ & - 6.1670 \times 10^{-5} t^3) S + (7.944 \times 10^{-2} + 1.6483 \times 10^{-2} t \\ & - 5.3009 \times 10^{-4} t^2) S^{3/2},\end{aligned}$$

$$A = A_w + (2.2838 \times 10^{-3} - 1.0981 \times 10^{-5} t - 1.6078 \times 10^{-6} t^2) S + 1.910\ 75 \times 10^{-4} S^{3/2}$$

$$B = B_w + (-9.9348 \times 10^{-7} + 2.0816 \times 10^{-8} t + 9.1697 \times 10^{-10} t^2) S$$

the pure water terms K_w , A_w and B_w of the secant bulk modulus are given by

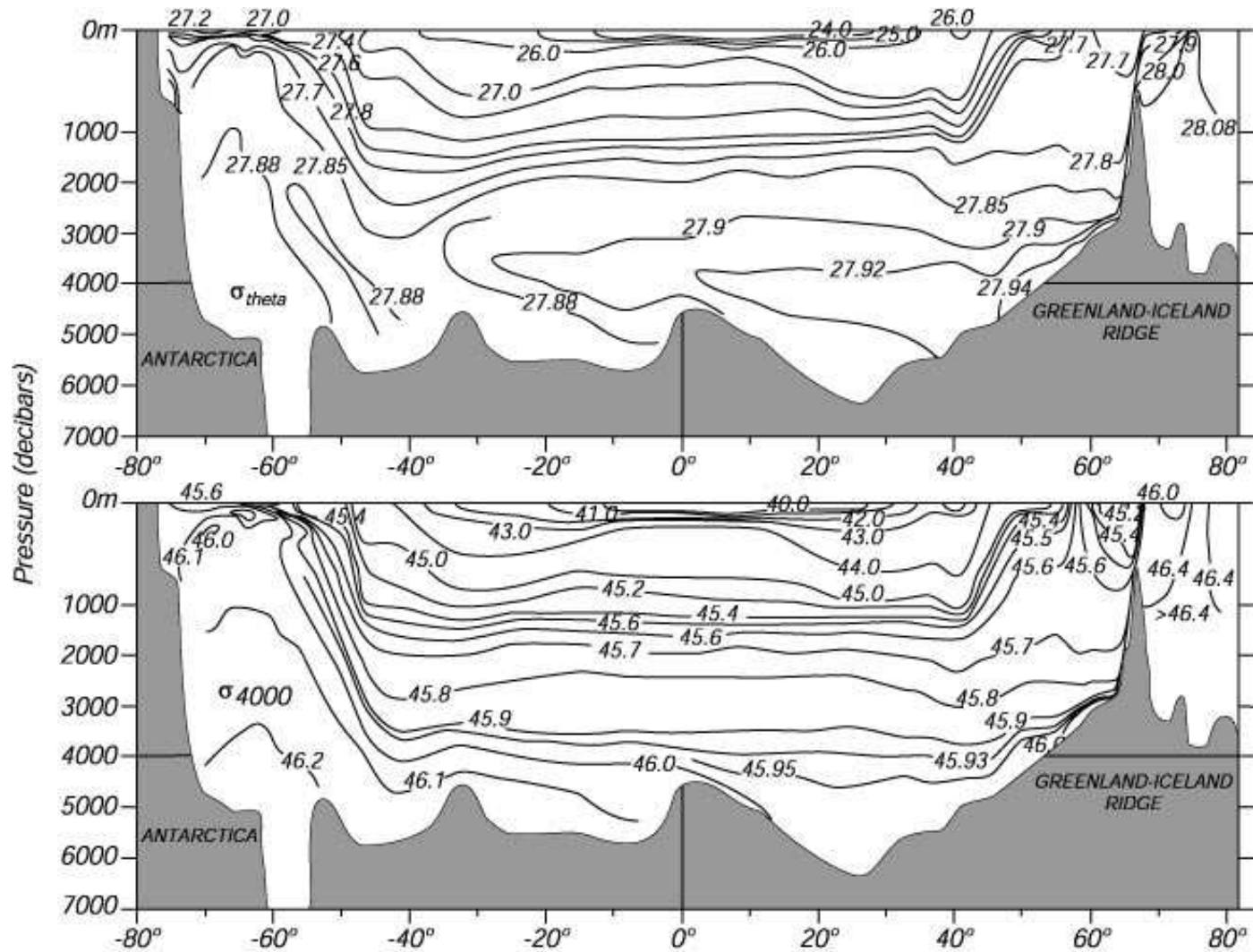
$$\begin{aligned}K_w = & 19\ 652.21 + 148.4206 t - 2.327\ 105 t^2 + 1.360\ 477 \times 10^{-2} t^3 \\ & - 5.155\ 288 \times 10^{-5} t^4\end{aligned}$$

$$\begin{aligned}A_w = & 3.239\ 908 + 1.437\ 13 \times 10^{-3} t + 1.160\ 92 \times 10^{-4} t^2 \\ & - 5.779\ 05 \times 10^{-7} t^3\end{aligned}$$

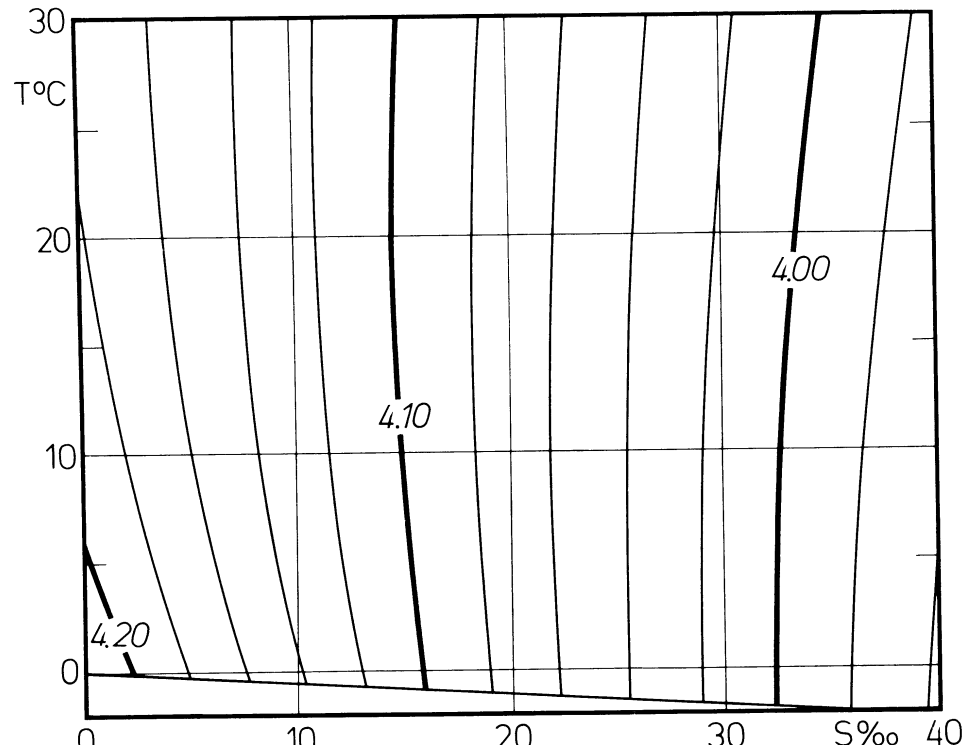
$$B_w = 8.509\ 35 \times 10^{-5} - 6.122\ 93 \times 10^{-6} t + 5.2787 \times 10^{-8} t^2$$

The high pressure International Equation of State of Seawater, 1980 is valid for practical salinity from 0 to 42, temperature from -2 to 40°C and applied pressure from 0 to 1000 bars.

Thermobaricity ; σ_θ VS σ_{4000}



Cappacité de Chaleur Massique



Sea Ice



Sea Ice



Sea Ice

